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Organic Preparations and Procedures International

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t902189982>

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To cite this Article Zisman, Stan A. , Berlin, K. Darrell and Scherlag, B. J.(1990) 'THE PREPARATION OF AMIDE DERIVATIVES OF 3-AZABICYCLO-[3.3.1]NONANES AS NEW POTENTIAL ANTIARRHYTHMIC AGENTS', *Organic Preparations and Procedures International*, 22: 2, 255 – 264

To link to this Article: DOI: 10.1080/00304949009458202

URL: <http://dx.doi.org/10.1080/00304949009458202>

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THE PREPARATION OF AMIDE DERIVATIVES OF 3-AZABICYCLO- [3.3.1]NONANES AS NEW POTENTIAL ANTIARRHYTHMIC AGENTS

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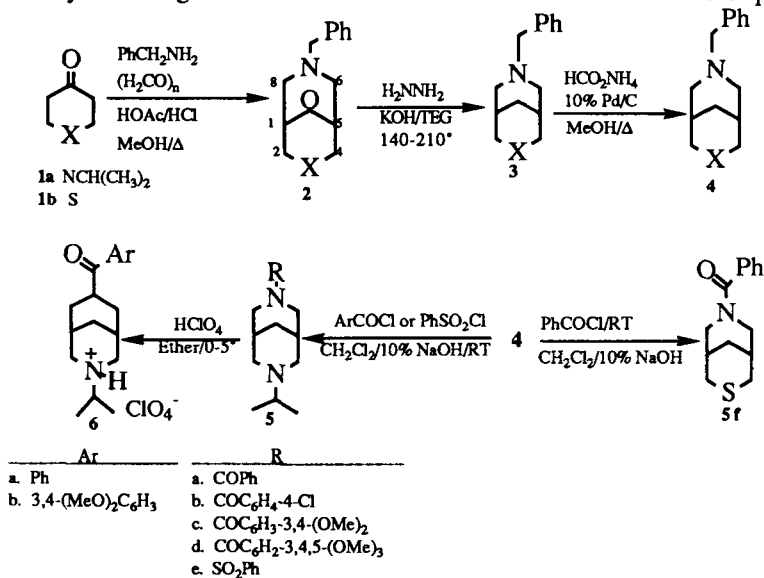
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Members of the 3-azabicyclo[3.3.1]nonanes¹⁻⁷ and certain derivatives possess promising pharmacological properties as supported by a variety of assays. Several compounds have displayed superb antiarrhythmic action in animal models and slight alterations in structure effect significant changes in the observed activity.²⁻⁵ A few isolated amide derivatives of this family have reported^{3,4} antiarrhythmic activity also. We record herein a general synthesis of certain amide derivatives which not only exhibit antiarrhythmic characteristics but are also potential metabolites (which might be derived as *in vivo* degradation products) of members of the 3-azabicyclo[3.3.1]nonanes.⁵

A double Mannich condensation⁸ of either 1-isopropyl-4-piperidinone (**1a**) or 4-thianone (**1b**) was utilized in the synthesis of the 3-hetera-7-azabicyclo[3.3.1]nonan-9-ones **2a** and **2b**, respectively. Condensation of **1a** or **1b**, benzylamine, paraformaldehyde, and acetic acid/HCl in methanol gave ketones **2a** and **2b**. It was discovered that the ratio of acetic acid to HCl was critical in that the isolated yield of **2a**, for example, was increased from 20-50% to a consistent 56-57%. This pH dependence suggests that iminium ion intermediates formed in the reaction prior to cyclization to the product require a high acidity medium, and/or the reaction kinetics are altered in such a fashion to favor formation of the bicyclic ketones **2**. Wolff-Kishner reduction⁹ of ketones **2** in the presence of hydrazine, KOH, and triethylene glycol at elevated temperature (140-210°) gave the amines **3** as oils (95%) which were characterized and used without further purification. Debenzylation of amines **3a** and **3b** was done in a boiling mixture of ammonium

formate, 10% Pd/C (catalytic), and methanol and gave the secondary amines **4a** (93%) and **4b** (57%), respectively. Although **4a** was isolated as an oil and used without further purification,



amine **4b** required chromatography and was isolated as a viscous gum. The use of ammonium formate as a hydrogen source (especially in debenzoylation) has been recently examined and has proven to be a mild method for effecting ArCH₂-N cleavage.¹⁰ However, it was surprising that hindered benzyl groups in **3** could be cleaved so readily. Amines **4** could then be acylated using a modified Schotten-Baumann procedure in which the amine was stirred at RT with the appropriate acylation agent in a biphasic mixture of CH₂Cl₂/10% NaOH to give the crude amides **5**. Chromatography (neutral alumina) afforded **5a** as an oil (82.4%) while amides **5b-d** and **5f** were isolated (53-80%) as solids. Compounds **5a** and **5c** were then derivatized by conversion to the respective hydroperchlorate salts **6a** (91%) and **6b** (69%). Similarly, amine **4a** was converted to the sulfonamide **5e**.

Amide rotational barriers exhibited by similar benzamides have been found to be approximately 14-15 kcal/mol from ¹H and ¹³C NMR spectral analyses in DCCl₃.^{11,12} Thus, in amides **5**, nonequivalence is observed for the CH₃ carbons, C(1) and C(5), C(2) and C(4), and C(6) and C(8). Spectral data for salts **6a** and **6b** were accumulated at 80° in DMSO-*d*₆ solution (a higher temperature was necessary to obtain coalesced, intensified signals since the spectra at

PREPARATION OF AMIDE DERIVATIVES OF THE 3-AZABICYCLO[3.3.1]NONANES

room temperature displayed broadened, unresolved signals). This suggests that the rotational barrier for salts **6** is much lower than for the unprotonated amides **5**.⁵ In summary, we report a convenient synthesis of amides of 3-azabicyclo[3.3.1]nonanes.

EXPERIMENTAL SECTION

All ¹H, ¹³C, and ¹⁵N spectral data were obtained on a Varian XL-300 NMR spectrometer operating at 299.94, 75.43, and 30.41 MHz, respectively. Chemical shifts for ¹H and ¹³C NMR spectra were recorded in δ or ppm values downfield from TMS [(CH₃)₄Si], while ¹⁵N NMR signals were reported in ppm downfield from NH₃ (*liquid*, 0 ppm) using 8 M ¹⁵NH₄NO₃ (19.73 ppm) as an external reference. IR spectra were acquired on a Perkin Elmer 681 IR spectrometer. Melting points, which were uncorrected, were recorded on a Thomas-Hoover capillary melting point apparatus. Elemental analyses were performed by Galbraith Laboratories, Inc., Knoxville, Tennessee. Syntheses were executed, unless otherwise indicated, under an atmosphere of N₂ with magnetic stirring. All reagents were obtained commercially and used without further purification except for 4-thianone (**1b**) [prepared by known methods¹³ and sublimed (45°/0.05 mm Hg)], 1-isopropyl-4-piperidinone (**1a**) (distilled: bp 38-41°/0.2 mm Hg), ammonium formate [recrystallized from methanol then dried (overnight, 0.2 mm Hg) in a desiccator prior to use (mp 118-120°)], and methanol (deoxygenated prior to use). Alumina (Merck, neutral, 70-230 mesh) and silica gel (Davison Chemical, "Davisil 62", 60-200 mesh) were employed in chromatographic separations with reagent grade solvents as eluants. *Caution: Although no difficulties were found with the hydroperchlorates cited herein, all work should be done in a hood and with extreme care.*

7-Benzyl-3-isopropyl-3,7-diazabicyclo[3.3.1]nonan-9-one (2a).-In standard equipment was placed a mixture of benzylamine (10.71 g, 100 mmol), HCl (37%, 9.86 g, 100 mmol), glacial acetic acid (3.0 g, 50 mmol), and paraformaldehyde (6.31 g, 210 mmol) in deoxygenated CH₃OH (100 mL) which was stirred at reflux (15 min). A solution of *N*-isopropyl-4-piperidinone (**1a**, 14.12 g, 100 mmol) and glacial acetic acid (6.0 g, 100 mmol) in CH₃OH (100 mL) was then added dropwise to the mixture (30 min), followed by stirring at reflux (18.5 hrs). Concentration of the solution gave an oil which was redissolved in H₂O (100 mL). An ether extract (100 mL) of this acidic solution was discarded. Basicification (pH~ 3, 10% NaOH) of the water layer produced a milky suspension which was extracted (ether, 4 x 60 mL). The combined extracts were dried (Na₂SO₄), filtered, and concentrated to a viscous red oil, which, on distillation (175-185°/10⁻⁵ mm Hg), afforded a light yellow oil (15.6 g, 57.2%) which solidified when refrigerated; mp. 46-47.5°. Recrystallization (pentane) gave an analytical sample of **2a**, mp. 49-50°. IR (KBr): 3035, 2975, 2900, 2820, 1745 (C=O) cm⁻¹; ¹H NMR

(DCCl₃): δ 1.02 [d, 6 H, CH₃], 2.58 [bs, 2 H, H(1,5)], 2.87 [m, 5 H, ring protons and CH(CH₃)₂], 3.03 (dd, 4 H, ring protons), 3.53 (s, 2 H, ArCH₂), 7.30 (m, 5 H, Ar-H); ¹³C NMR (DCCl₃): ppm 18.25 (CH₃), 46.93 [C(1,5)], 53.41 [CH(CH₃)₂], 53.71 [C(2,4)], 58.07 [C(6,8)], 61.25 (ArCH₂), 127.09, 128.25, 128.69, 138.67 (ArC), 215.20 (C=O); ¹⁵N NMR (DCCl₃): ppm 39.25 [N(7)], 40.80 [N(3)].

Anal. Calcd. for C₁₇H₂₄N₂O: C, 74.96; H, 8.88; N, 10.28.

Found: C, 75.18; H, 8.61; N, 10.24.

7-Benzyl-3-thia-7-azabicyclo[3.3.1]nonan-9-one (2b).-Similar to **2a**, a mixture containing benzylamine (0.43 g, 4 mmol), paraformaldehyde (0.96 g, 32 mmol), CH₃OH (15 mL), HCl (37%, 0.20 g, 2 mmol), and glacial acetic acid (0.36 g, 6 mmol) was treated with 4-thianone (**1b**, 0.47 g, 4 mmol) and heated at reflux (6 hrs). Evaporation of the solvent gave a red oil which was diluted (H₂O, 30 mL), and the mixture was extracted (ether, 2 x 40 mL), the latter being discarded. Basification (pH~2, NaOH pellets 97%, 0.26 g, 6 mmol) of the aqueous layer produced a yellow suspension which was extracted (ether, 4 x 30 mL). Combined extracts were dried (Na₂SO₄), filtered, and concentrated to a yellow solid. This solid was digested (Skelly B, 300 mL, bp 60-68°) for 30 min, and the supernatant was decanted. Evaporation of the solvent gave **2b** which was then heated *in vacuo* (110°/0.1 mm) in a sublimation unit to give ketone **2b** (0.4 g, 40.4%); mp. 94.5-95.5° (lit.² mp. 91-93°).

7-Benzyl-3-isopropyl-3,7-diazabicyclo[3.3.1]nonane (3a).-To a mixture of KOH pellets (85%, 13.6 g, 206 mmol) and ketone **2a** (7.0 g, 25.7 mmol) in triethylene glycol (120 mL) was added hydrazine (95%, 3.47 g, 103 mmol) in equipment with a lower take-off condenser and a jacketed flask (boiling tetralin in the jacket gave a temperature of 200-210°). After 4 hrs at reflux, the solution was cooled to RT and diluted (chilled water, 150 mL). Combined extracts (ether, 4 x 60 mL) of the suspension were washed with 10% NaOH (60 mL) and saturated NaCl (60 mL), dried (Na₂SO₄), filtered, and concentrated to a yellow oil (6.33 g, 95.3%). IR analysis showed no carbonyl and thus this oil **3a** was used directly.

7-Benzyl-3-thia-7-azabicyclo[3.3.1]nonane (3b).-To a mixture of KOH pellets (85%, 3.20 g, 48.5 mmol) and ketone **2b** (1.0 g, 4.04 mmol) in triethylene glycol (25 mL) was added,

PREPARATION OF AMIDE DERIVATIVES OF THE 3-AZABICYCLO[3.3.1]NONANES

hydrazine (95%, 1.36 g, 40.4 mmol) in a jacketed flask (*o*-xylene, bp 140-150°). Following the above procedure gave 0.90 g (95.4%) of a light yellow oil **3b** which displayed no C=O stretch in the IR spectrum and was used without further purification.

3-Isopropyl-3,7-diazabicyclo[3.3.1]nonane (4a).-In standard equipment was stirred a mixture of amine **3a** (5.53 g, 21.4 mmol) and 10% Pd/C (0.64 g, 30 mg/mmol of amine) in CH₃OH (80 mL). To this was added anhydrous HCO₂NH₄ (3.37 g, 53.5 mmol). Stirring the mixture at reflux (30 min), cooling the new mixture to RT, and filtering was followed by concentration to give a viscous oil. The oil was dissolved in H₂O (80 mL) and the pH was adjusted (~12, 10% NaOH). Combined extracts (CCl₄, 4 x 40 mL) of the aqueous solution were dried (Na₂SO₄), filtered, and concentrated to give yellow oil **4a** (3.35 g, 93.0%) which was used directly. IR (film): 3315 (N-H), 2965, 2900, 2850, 2790, 2760, 2725 cm⁻¹; ¹H NMR (DCCl₃): δ 1.01 [d, 6 H, CH₃, J = 6.7 Hz], 1.60-1.67 [m, 3 H, H(1,5) and H(9)], 1.79-1.84 [m, 1 H, H(9)], 2.53-2.59 [m, 3 H, H(6,8)ax and CH(CH₃)₂], 2.90-3.06 [m, 4 H, H(6,8)eq, H(2,4)ax and H(2,4)eq], 3.56 (bs, 1 H, NH); ¹³C NMR (DCCl₃): ppm 18.12 (CH₃), 30.04 [C(1,5)], 33.62 [C(9)], 52.86 [C(6,8)], 54.59 [CH(CH₃)₂], 54.65 [C(2,4)].

3-Thia-7-azabicyclo[3.3.1]nonane (4b).-In standard equipment was placed amine **3b** (0.90 g, 3.86 mmol) and 10% Pd/C (0.90 g) in anhydrous CH₃OH (25 mL), and anhydrous HCO₂NH₄ (1.11 g, 17.1 mmol) was added. With stirring, the mixture was brought to reflux (30 min), filtered, and concentrated to a gummy oil which was dissolved in CH₂Cl₂ (15 mL) and filtered. The filtrate was concentrated and placed in a diffusion chamber of ether. The mother liquor was decanted from the crude, oily **4b**. Chromatography of the oil was effected with CH₃OH/CH₂Cl₂ (300 mL of 10% CH₃OH/CH₂Cl₂, 50 mL of 20% CH₃OH/CH₂Cl₂, 100 mL of 50% CH₃OH/CH₂Cl₂, and 100 mL of CH₃OH) on silica gel. This afforded 0.32 g (56.9%) of amine **4b** (R_f = 0.11, 10% CH₃OH/CH₂Cl₂) as a light, gummy solid (used without further purification). ¹H NMR (DCCl₃): δ 1.84, 2.04 [two bd, 2 H, H(9)], 2.31 [bs, 2 H, H(1,5)], 2.80 [bd, 2 H, H(2,4)ax, J = 12.3 Hz], 3.20 [bd, 2 H, H(2,4)eq, J = 13.7 Hz], 3.45 [m, 2 H, H(6,8)ax], 3.73 [bd, 2 H, H(6,8)eq, J = 13.2 Hz], 7.59 (bs, 1 H, NH); ¹³C NMR (DCCl₃): ppm 24.88 [d, C(1,5)], 29.79 [t, C(9)], 32.17 [t, C(2,4)], 47.8 [t, C(6,8)].

3-Benzoyl-7-isopropyl-3,7-diazabicyclo[3.3.1]nonane (5a).-The following experiment will illustrate the general procedure used for most amides. In standard equipment was placed a solution of the amine **4a** (1.14 g, 6.77 mmol) and NaOH (10%, 6.80 g, 16.9 mmol) in CH₂Cl₂ (15 mL). Dropwise addition (15 min) of a solution of benzoyl chloride (1.05 g, 7.45 mmol) in CH₂Cl₂ (5 mL) was followed by stirring (2.75 hrs). Addition of H₂O (30 mL) and extraction (CH₂Cl₂, 3 x 25 mL) followed. The combined extracts were dried (Na₂SO₄), filtered, and concentrated to a yellow oil. Chromatography of the oil (neutral alumina, EtOAc) gave fractions (R_f = 0.70) which were combined and concentrated to give 1.52 g (82.4%) of amide **5a** as an oil (used directly). IR (film): 3035, 2970, 2925, 2865, 2805, 2780, 2750, 1635 (C=O) cm⁻¹; ¹H NMR (DCCl₃): δ 0.96 [d, 3 H, CH₃, J = 6.4 Hz], 1.07 [d, 3 H, CH₃, J = 6.6 Hz], 1.65-1.78, [m, 3 H, H(5) and H(9)], 1.97 [bs, 1 H, H(1)], 2.41 [d, 1 H, H(4)_{ax}, J = 10.3 Hz], 2.50 [d, 1 H, H(6)_{ax}, J = 11.0 Hz], 2.62 [m, 1 H, CH(CH₃)₂, J = 6.5 Hz], 2.72 [d, 1 H, H(6)_{eq}, J = 10.6 Hz], 3.03-3.07 [m, 2 H, H(2)_{ax} and H(4)_{eq}], 3.30 [d, 1 H, H(8)_{ax}, J = 13.2 Hz], 3.74 [d, 1 H, H(8)_{eq}, J = 12.8 Hz], 4.77 [d, 1 H, H(2)_{eq}, J = 13.9 Hz], 7.28-7.41 (m, 5 H, Ar-H); ¹³C NMR (DCCl₃): ppm 16.30 [CH₃], 19.33 [CH₃], 29.06 [C(1)], 29.76 [C(5)], 32.29 [C(9)], 46.55 [C(2)], 52.19 [C(4)], 52.62 [C(8)], 54.34 [CH(CH₃)₂], 54.75 [C(6)], 126.75, 128.24, 128.67, 137.75 (Ar-C), 170.09 (C=O).

3-(4-Chlorobenzoyl)-7-isopropyl-3,7-diazabicyclo[3.3.1]nonane (5b).-A mixture of amine **4a** (0.60 g, 3.57 mmol) in CH₂Cl₂ (5 mL) and 10% NaOH (3.58 g, 8.93 mmol) was treated (15 min) with 4-chlorobenzoyl chloride (0.69 g, 3.92 mmol) in CH₂Cl₂ (5 mL). The usual procedure gave 0.86 g (80.4%) of **5b**; mp. 97-98°. IR (KBr): 2965, 2935, 2865, 2800, 2770 (C-H), 1630 (C=O) cm⁻¹; ¹H NMR (DCCl₃): δ 0.95 [d, 3 H, CH₃, J = 6.5 Hz], 1.05 [d, 3 H, CH₃, J = 6.4 Hz], 1.63-1.75 [m, 3 H, H(5) and H(9)], 1.97 [bs, 1 H, H(1)], 2.41 [bd, 1 H, H(4)_{ax}, J = 10.6 Hz], 2.50 [bd, 1 H, H(6)_{ax}, J = 11.2 Hz], 2.59 [heptet, 1 H, CH(CH₃)₂, J = 6.5 Hz], 2.71 [bd, 1 H, H(6)_{eq}, J = 11.0 Hz], 3.03-3.06 [m, 2 H, H(2)_{ax} and H(4)_{eq}], 3.31 [bd, 1 H, H(8)_{ax}, J = 12.8 Hz], 3.71 [bd, 1 H, H(8)_{eq}, J = 13.1 Hz], 4.77 [bd, 1 H, H(2)_{eq}, J = 13.2 Hz], 7.27-7.37 (m, 4 H, Ar-H); ¹³C NMR (DCCl₃): ppm 16.37, 19.35 [CH₃], 29.07 [C(1)], 29.80 [C(5)], 32.29 [C(9)], 46.68 [C(2)], 52.22 [C(4)], 52.56 [C(8)],

PREPARATION OF AMIDE DERIVATIVES OF THE 3-AZABICYCLO[3.3.1]NONANES

54.38 [CH(CH₃)₂], 54.79 [C(6)], 128.35, 128.51, 134.67, 136.11 (Ar-C), 169.03 (C=O).

Anal. Calcd. for C₁₇H₂₃ClN₂O: C, 66.55; H, 7.56. Found: C, 66.45; H, 7.71

3-(3,4-Dimethoxybenzoyl)-7-isopropyl-3,7-diazabicyclo[3.3.1]nonane (5c).-A mixture of amine **4a** (0.60 g, 3.57 mmol) in CH₂Cl₂ (5 mL) and 10% NaOH (3.58 g, 8.93 mmol) was treated (15 min) with 3,4-dimethoxybenzoyl chloride (0.80 g, 3.92 mmol) in CH₂Cl₂ (10 mL).

The usual procedure gave 0.87 g (73.1%) of off-white solid **5c**; mp. 67.5-69.5°. IR (KBr): 3055, 2950, 2915, 2845, 2820, 2770, 2750, 2710, 1625 (C=O) cm⁻¹; ¹H NMR (DCCl₃): δ 0.96 [d, 3 H, CH₃, J = 6.4 Hz], 1.06 [d, 3 H, CH₃, J = 6.5 Hz], 1.62-1.75 [m, 3 H, H(5) and H(9)], 1.96 [bs, 1 H, H(1)], 2.43 [bd, 1 H, H(4)ax, J = 9.7 Hz], 2.51 [bd, 1 H, H(6)ax, J = 10.5 Hz], 2.62 [heptet, 1 H, CH(CH₃)₂, J = 6.4 Hz], 2.74 [bd, 1 H, H(6)cq, J = 9.9 Hz], 3.00-3.09 [m, 2 H, H(4)eq and H(2)ax], 3.32 [bd, 1 H, H(8)ax, J = 13.2 Hz], 3.83-3.94 [m, 7 H, H(8)eq and OCH₃], 4.77 [bd, 1 H, H(2)eq, J = 13.3 Hz], 6.84-6.94 (m, 3 H, Ar-H); ¹³C NMR (DCCl₃): ppm 16.46, 19.15 [CH₃], 29.13 [C(1)], 29.86 [C(5)], 32.36 [C(9)], 46.71 [C(2)], 52.25 [C(4)], 52.65 [C(8)], 54.35 [CH(CH₃)₂], 54.68 [C(6)], 55.88, 55.93 (OCH₃), 110.50, 119.64, 130.23, 148.78, 149.39 (Ar-C), 169.90 (C=O).

Anal: Calcd. for C₁₉H₂₈N₂O₃: C, 68.65; H, 8.49. Found: C, 68.58; H, 8.47

7-Isopropyl-3-(3,4,5-trimethoxybenzoyl)-3,7-diazabicyclo[3.3.1]nonane (5d).-A mixture of amine **4a** (0.60 g, 3.57 mmol) in CH₂Cl₂ (5 mL) and 10% NaOH (3.58 g, 8.93 mmol) was treated with 3,4,5-trimethoxybenzoyl chloride (0.92 g, 3.92 mmol) in CH₂Cl₂ (5 mL). This gave 1.02 g (79.1%) of off-white solid **5d**; mp. 67.5-69.5°. IR (KBr): 3055, 2985, 2955, 2910, 2890, 2780 (C-H), 1620 (C=O) cm⁻¹; ¹H NMR (DCCl₃): δ 0.96 [d, 3 H, CH₃, J = 6.5 Hz], 1.09 [d, 3 H, CH₃, J = 6.7 Hz], 1.64-1.79 [m, 3 H, H(5) and H(9)], 2.05 [bs, 1 H, H(1)], 2.44 [bd, 1 H, H(4)ax, J = 10.6 Hz], 2.57 [bd, 1 H, H(6)ax, J = 10.8 Hz], 2.66 [heptet, 1 H, CH(CH₃)₂, J = 6.6 Hz], 2.71 [bd, 1 H, H(6)eq, J = 11.0 Hz], 3.02-3.07 [m, 2 H, H(4)eq and H(2)ax], 3.31 [bd, 1 H, H(8)ax, J = 13.2 Hz], 3.80-3.92 [m, 10 H, H(8)eq and OCH₃], 4.77 [bd, 1 H, H(2)eq, J = 13.5 Hz], 7.29 (s, 2 H, Ar-H); ¹³C NMR (DCCl₃): ppm 15.87, 19.42 (CH₃), 29.02 [C(1)], 29.78 [C(5)], 32.35 [C(9)], 46.64 [C(2)], 51.73 [C(4)], 52.48 [C(8)], 54.39 [CH(CH₃)₂], 54.95 [C(6)], 56.13, 60.86 (OCH₃), 103.83,

133.32, 133.21, 138.22, 153.21 (Ar-C), 169.66 (C=O).

Anal. Calcd. for $C_{20}H_{30}N_2O_4$: C, 66.27; H, 8.34. Found: C, 66.04; H, 8.32

3-Benzenesulfonyl-7-isopropyl-3,7-diazabicyclo[3.3.1]nonane (5e).-A stirred, ice cold (5°) mixture of amine **4a** (1.03 g, 6.12 mmol) and NaOH pellets (97%, 0.76 g, 18.4 mmol) in H_2O (7 mL) and CH_2Cl_2 (5 mL) was treated (30 min) with benzenesulfonyl chloride (2.16 g, 12.2 mmol) in CH_2Cl_2 (5 mL). Stirring (17.5 hrs) at RT gave a mixture which was partitioned between H_2O (30 mL) and CH_2Cl_2 (30 mL) followed by basification (pH~12, 10% NaOH) of the aqueous phase. Aqueous extracts (CH_2Cl_2 , 3 x 30 mL) and the initial organic layer were combined and washed [10% NaOH (30 mL); saturated NaCl (30 mL)], dried (Na_2SO_4), filtered, and concentrated to a viscous oil. Chromatography (silica gel, 10% CH_3OH/CH_2Cl_2) gave fractions ($R_f = 0.44$) which were concentrated and rechromatographed (neutral alumina, EtOAc). Fractions ($R_f = 0.53$) were concentrated, and rechromatographed (silica gel, 5% CH_3OH/CH_2Cl_2). Fractions ($R_f = 0.34$) were concentrated to give 0.54 g (28.6%) of white solid **5e**; mp. $85.5-86.5^\circ$. IR (KBr): 3060, 2960, 2910, 2890, 2865, 2820, 1585, 1340, 1170 ($S=O$) cm^{-1} ; 1H NMR (DMSO- d_6): δ 0.88 [d, 6 H, CH_3 , $J = 6.5$ Hz], 1.40 [bs, 2 H, H(9)], 1.94 [bs, 2 H, H(1,5)], 2.35 [bd, 2 H, H(6,8)ax, $J = 10.3$ Hz], 2.53 [heptet, 1 H, $CH(CH_3)_2$, $J = 6.5$ Hz], 2.69 [bd, 2 H, H(6,8)cq, $J = 10.3$ Hz], 2.89 [dd, 2 H, H(2,4)ax, $J = 11.2$ Hz, $J' = 4.5$ Hz], 3.36 [d, 2 H, H(2,4)eq, $J = 10.9$ Hz], 7.58-7.75 (m, 5 H, Ar-H); ^{13}C NMR (DMSO- d_6): ppm 17.57 (CH_3), 27.39 [C(1,5)], 28.95 [C(9)], 48.88 [C(2,4)], 52.66 [C(6,8)], 53.42 [$CH(CH_3)_2$], 126.90, 129.01, 132.36, 136.79 (ArC).

Anal. Calcd. for $C_{16}H_{24}N_2O_2S$: C, 62.31; H, 7.85. Found: C, 62.48; H, 7.69.

7-Benzoyl-3-thia-7-azabicyclo[3.3.1]nonane (5f).-In the usual setup, a chilled (5°) solution of NaOH pellets (0.1 g, 2.38 mmol) in H_2O (1.7 mL) and amine **4a** (0.17 g, 1.19 mmol) in CH_2Cl_2 (1 mL) was treated (5 min) with benzoyl chloride (0.2 g, 1.43 mmol). Chromatography of the oil (alumina, EtOAc), afforded amide **5f** ($R_f = 0.47$) as white crystals (157 mg, 53.3%); mp. $95-96^\circ$. IR (KBr): 3065, 3045, 3000, 2985, 2940, 2910, 2855, 2835, 1635 ($C=O$); 1H NMR ($DCCl_3$): δ 1.78-1.93 [m, 3 H, H(9) and H(1)], 2.15 [bs, 1 H, H(5)], 2.39 [d, 1 H, H(4)ax, $J = 13.9$ Hz], 2.77 [d, 1 H, H(6)ax, $J = 12.3$ Hz], 3.12-3.21 [m, 3 H,

PREPARATION OF AMIDE DERIVATIVES OF THE 3-AZABICYCLO[3.3.1]NONANES

H(4)eq and H(6)eq], 3.41 [d, 1 H, H(2)ax, $J = 12.8$ Hz], 3.89 [d, 1 H, H(2)eq, $J = 13.4$ Hz], 4.98 [d, 1 H, H(8)eq, $J = 13.1$ Hz], 7.38-7.44 (m, 5 H, Ar-H); ^{13}C NMR (DCCl_3): ppm 26.53 [C(1)], 26.87 [C(5)], 31.73 [C(2)], 31.78 [C(9)], 32.34 [C(4)], 46.07 [C(8)], 52.12 [C(6)], 126.46, 128.41, 128.83, 137.35 (Ar-C), 170.38 (C=O).

Anal. Calcd. for $\text{C}_{14}\text{H}_{17}\text{NOS}$: C, 67.98; H, 6.93. Found: C, 68.01; H, 7.07

3-Benzoyl-7-isopropyl-3,7-diazabicyclo[3.3.1]nonane Hydroperchlorate (6a). -A chilled (5°), stirred solution of the amide **5a** (1.52 g, 5.58 mmol) in ether (60 mL) was treated (10 min) with a solution of HClO_4 (60%, 1.17 g, 6.98 mmol) followed by stirring (10 min). Filtration gave salt **6a** which was washed (dry ether, 50 mL), stirred [hot CH_3OH (30 mL)] and refrigerated. Filtration and drying afforded 1.90 g (91.3%) of **6a**; mp. 226-227° (dec): IR (KBr): 3150 (NH), 2990, 2960, 2935, 2920, 2885, 1635 (C=O) cm^{-1} , 1100 (Cl-O); ^{14}H NMR [$\text{DMSO}-d_6$]: δ 1.55 [d, 6 H, CH_3 , $J = 6.6$ Hz], 1.97 [bd, 1 H, H(9), $J = 13.0$ Hz], 2.18 [bd, 1 H, H(9), $J = 13.2$ Hz], 2.51 [bs, 2 H, H(1,5)], 3.30 [bd, 2 H, H(6,8)ax, $J = 13.2$ Hz], 3.65 [m, 2 H, H(2,4)ax], 3.83 [h, 1 H, $\text{CH}(\text{CH}_3)_2$, $J = 6.8$ Hz], 3.94 [bd, 2 H, H(6,8)eq, $J = 12.3$ Hz], 4.23 [bd, 2 H, H(2,4)eq, $J = 13.2$ Hz], 7.45-7.50 (m, 5 H, Ar-H), 7.85 (bs, 1 H, N-H); ^{13}C NMR ($\text{DMSO}-d_6$, 80°): ppm 16.34 [CH_3], 26.69 [C(1,5)], 27.62 [C(9)], 48.80 [C(2,4)], 52.31 [C(6,8)], 59.91 [$\text{CH}(\text{CH}_3)_2$], 127.05, 128.30, 129.40, 136.40 (Ar-C), 172.86 (C=O).

Anal. Calcd. for $\text{C}_{17}\text{H}_{25}\text{ClN}_2\text{O}_5$: C, 54.76; H, 6.76. Found: C, 54.43; H, 6.78

3-(3,4-Dimethoxybenzoyl)-7-isopropyl-3,7-diazabicyclo[3.3.1]nonane Hydroperchlorate (6b)- Amide **5c** (0.30 g, 0.90 mmol) in ether (30 mL) was treated with a solution of HClO_4 (60%, 0.18 g, 1.08 mmol) in $(\text{H}_3\text{C})_2\text{CHOH}$ (1 mL). The usual procedure gave 0.27 g (69.2%) of white solid **6b**; mp. 235-236° (dec). IR (KBr): 3130 (NH), 3010, 2975, 2945, 2920, 1635 (C=O), 1095 (Cl-O) cm^{-1} ; ^1H NMR ($\text{DMSO}-d_6$, 80°): δ 1.33 [d, 6 H, CH_3 , $J = 6.7$ Hz], 1.74 [bd, 1 H, H(9), $J = 12.8$ Hz], 1.91 [bd, 1 H, H(9), $J = 13.2$ Hz], 2.27 [bs, 2 H, H(1,5)], 3.12 [bd, 2 H, H(6,8)ax, $J = 13.7$ Hz], 3.19-3.28 [m, 2 H, H(2,4)ax], 3.42-3.56 [m, 3 H, H(6,8)eq and $\text{CH}(\text{CH}_3)_2$], 3.78, 3.81 (two s, 6 H, OCH_3), 3.97 [bd, 2 H, H(2,4)eq, $J = 13.6$ Hz], 6.94-7.03 (m, 3 H, Ar-H), 7.81 (bs, 1 H, N-H); ^{13}C NMR ($\text{DMSO}-d_6$, 80°): ppm 16.34 (CH_3), 26.81 [C(1,5)], 27.80 [C(9)], 49.06 [C(2,4)], 52.40 [C(6,8)], 55.88, 55.92 (OCH_3),

59.97 [CH(CH₃)₂], 111.85, 112.10, 120.44, 128.68, 148.74, 150.26 (Ar-C), 172.97 (C=O).

Anal. Calcd. for C₁₉H₂₉ClN₂O₇: C, 52.72; H, 6.75. Found: C, 52.35; H, 6.77

Acknowledgment.-We gratefully acknowledge a Fellowship from the Presidential Water Resources Fellowship Program to one of us (SAZ). We also acknowledge partial support by the College of Arts and Sciences in the form of salary (KDB). Grateful acknowledgment is given to the National Science Foundation (grant DMB-8603864) and to OCAST (grant 1506) for partial support to upgrade the XL-300 NMR spectrometer.

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(Received July 24, 1989; in revised form November 15, 1989)